

Marine productivity estimates from continuous O₂/Ar ratio measurements by membrane inlet mass spectrometry

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[1] Dissolved oxygen/argon (O₂/Ar) ratios in the oceanic mixed layer are indicative of net community production (NCP) because O₂ and Ar share similar physical solubility properties, but only O₂ is biologically produced and consumed. We describe a membrane inlet mass spectrometer (MIMS) that allows continuous high-precision shipboard analysis of O₂/Ar ratios and eventually other gases, calibrated with discrete samples analyzed in the laboratory. We also present O₂/Ar data from the eastern equatorial Pacific. Short-term reproducibilities of 0.05% were achieved. Meridional gradients and small-scale phenomena were clearly resolved. O₂/Ar undersaturations around the equator reflect the interaction of biological and physical forcings. Mixed-layer NCP estimated from wind speed-gas exchange parameterizations was near zero north of 2.75°N, and about 12 mmol m⁻² d⁻¹ south of 6.75°S. Ar supersaturations, calculated from MIMS O₂/Ar measurements and accompanying O₂ concentration measurements, ranged from -0.8 to +3.0%. **Citation:** Kaiser, J., M. K. Reuer, B. Barnett, and M. L. Bender (2005), Marine productivity estimates from continuous O₂/Ar ratio measurements by membrane inlet mass spectrometry, *Geophys. Res. Lett.*, *32*, L19605, doi:10.1029/2005GL023459.

1. Introduction

[2] The dissolved oxygen (O₂) concentration of seawater varies because of fundamental physical and biological processes. These include photosynthesis and respiration, diffusive and bubble-mediated gas exchange, temperature and pressure changes, lateral mixing and vertical diffusion.

[3] The marine oxygen cycle is tightly linked to the carbon cycle via photosynthesis and respiration. In the absence of physical effects, dissolved O₂ constrains the balance of these two biological processes, i.e., net community production (NCP). O₂ can be used as a geochemical tracer that reflects carbon fluxes integrated over characteristic response times. Warming and bubble injection lead to O₂ supersaturation, posing a challenge to this approach.

[4] Craig and Hayward [1987] used oxygen/argon (O₂/Ar) ratios to separate O₂ supersaturations into a biological and a physical component. This method is based on the similar solubility characteristics of these two gases with respect to temperature and pressure changes as well as bubble-mediated gas exchange. One can define an O₂/Ar supersaturation, ΔO₂/Ar, as:

$$\Delta O_2/Ar = \frac{c(O_2)}{c(Ar)} \bigg/ \frac{c_{sat}(O_2)}{c_{sat}(Ar)} - 1$$

[5] ΔO₂/Ar essentially records the difference between photosynthetic O₂ production and respiration. c is the dissolved gas concentration (in mol m⁻³) and c_{sat} is the saturation concentration. c_{sat} is a function of temperature, pressure and salinity. This approach, in which discrete samples are collected at sea, stored, and analyzed in the lab, has been widely used in subsequent work [e.g., Emerson *et al.*, 1995; Hendricks *et al.*, 2004; Spitzer and Jenkins, 1989].

[6] Here, we describe a new method for continuous underway measurements of O₂/Ar by membrane-inlet mass spectrometry (MIMS), extending earlier oceanographic MIMS applications [Kana *et al.*, 1994; Tortell, 2005]. The measured ΔO₂/Ar values can be used in conjunction with suitable wind-speed gas-exchange parameterizations to calculate biologically induced air-sea O₂ fluxes and, where conditions are appropriate, NCP. The inferred NCP values represent rates integrated over the characteristic mixed layer gas exchange times (ratio of mixed layer thickness and piston velocity), typically between 10 and 30 days.

[7] In this paper, we show results of continuous O₂/Ar measurements in the eastern equatorial Pacific (EEP). The equatorial Pacific is a region of considerable interest because of its high-nutrient low chlorophyll character [Coale *et al.*, 1996], its role as oceanic CO₂ source [Takahashi *et al.*, 2002], and its strong coupling of biogeochemical variability to physical forcing. Here, we address a question related to this topic: How does net community production vary in the EEP between the nutrient-poor region north of the equatorial upwelling zone and the nutrient-rich region in the south? We present O₂ and O₂/Ar data in the region between 8°S and 12°N, 110°W and 95°W. The results exemplify the use of MIMS for oceanic productivity studies and its potential to provide large-scale constraints for satellite-based productivity estimates [e.g., Behrenfeld *et al.*, 2005].

2. Methods

[8] Continuous O₂/Ar ratio measurements were made by MIMS on board NOAA R/V Ron Brown during a cruise to service TAO (Tropical Atmosphere Ocean Project) [McPhaden *et al.*, 1998; McPhaden, 1995] buoys in the eastern equatorial Pacific in October/November 2003 (RB-03-09-TAO). Meridional and zonal transects were performed in the following order: 12°N to 8°S along 95°W, 95°W to 110°W along 8°S, 8°S to 8°N along 110°W, 110°W to 95°W along 8°N, 8°N to equator along 95°W. The ship's underway sampling system was used to pump water through a Teflon AF membrane (Random Technologies) mounted in a stainless steel vacuum manifold, which was

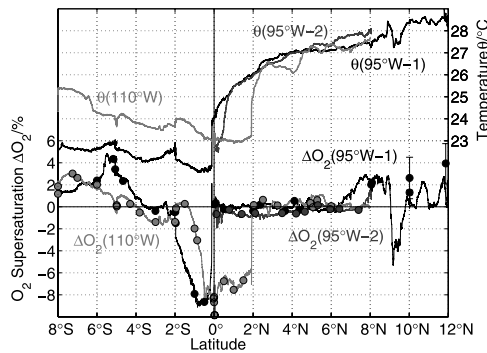


Figure 1. Sea surface temperature (θ) and oxygen supersaturation (ΔO_2) from optode measurements during the three meridional transects of RB-03-09-TAO at 95°W (1: 10/29/03 to 11/05/03; 2: 11/16/03 to 11/18/03, north to south) and 110°W (11/08/03 to 11/14/03, south to north). θ values are plotted as thicker lines than ΔO_2 values. Discrete Winkler O₂ measurements used for calibration of the optode measurements are indicated by filled circles. Error bars are smaller than the symbol size except for the first Winkler measurements between 10°N and 12°N, which were disregarded for the calibration of the optode. See color version of this figure in the HTML.

connected to a quadrupole mass spectrometer (*Pfeiffer Vacuum Prisma*). The intake of the underwater sampling system is located at the bow at a nominal depth of 5 m. The inner diameter of the membrane of 600 μm and the head pressure of the ship's continuous seawater pump (3 bar) determined the volumetric flow rate of 12 cm^3/min through the 10 cm-long membrane tubing. In order to reduce O₂/Ar variations due to temperature and water vapor pressure effects and to avoid degassing, the vacuum manifold with the membrane was held at a constant temperature of 20°C ($\geq 2^\circ\text{C}$ below the surface seawater temperature, SST) and the flight tube was in a thermally insulated box maintained at 40°C.

[9] We calibrated the O₂/Ar ratio measurements with discrete water samples taken from the same seawater outlet as used for the MIMS measurements. 300 cm^3 samples were drawn into pre-evacuated glass flasks poisoned with 7 mg HgCl₂ [Emerson *et al.*, 1999]. These samples were later analyzed with an isotope ratio mass spectrometer (IRMS, *Thermo Finnigan*) for their dissolved O₂/Ar ratios and the oxygen triple isotope composition relative to air [Hendricks *et al.*, 2004]. The precision of the IRMS O₂/Ar ratios was about 0.1%, based on analyses of duplicate samples. MIMS O₂/Ar ion current ratio measurements were made every 10 to 30 s and had a short-term stability of 0.05% (standard deviation s of 10 sample-mean; error estimates in the following are $\pm 1s$). The raw O₂/Ar ion current ratios were calibrated by linear regression against the IRMS-measured dissolved O₂/Ar ratios. The mean residual was $(0.0 \pm 0.3)\%$.

[10] In addition to the O₂/Ar ratios, O₂ concentrations were measured continuously with an optode (*Aanderaa*), calibrated by automatic Winkler titration of discrete water samples with potentiometric endpoint detection. The analytical precision of the Winkler method was better than 0.1%. Short-term (60 s) precision of the optode measure-

ments was 0.03%. Calibration was achieved by regression of the temperature-corrected optode readings against the Winkler results. The mean difference between calibrated optode and Winkler measurements of the O₂ supersaturation (ΔO_2) was $(0.0 \pm 0.3)\%$.

[11] The spatial sampling resolution at a cruise speed of 12 kn is 60–180 m for the MIMS and about 40 m for the optode measurements. Response times ($1/e$ -fold) are about 60 s for the MIMS and 30 s for the optode.

[12] Dissolved O₂ was also measured in surface water samples from Niskin bottles in order to assess whether any gas losses occurred from the water pumped from the seawater intake to the laboratory due to warming and potential outgassing or O₂ loss to the pipe walls. A mean ΔO_2 decrease of $(0.5 \pm 0.3)\%$ was recorded and corrected for in the results presented here. About one third of the ΔO_2 change can be explained with outgassing due to warming by about 0.08°C between seawater intake and outlet in the lab. We did not measure O₂/Ar ratios in Niskin samples on this cruise. On a 2004 cruise of R/V *Ka'Imimoana*, however, we found that the mean difference between Niskin and continuous measurements was negligible, with a similar change in ΔO_2 as observed for the RB-03-09-TAO cruise. We therefore do not apply any corrections to the $\Delta O_2/\text{Ar}$ values.

[13] ΔO_2 and $\Delta O_2/\text{Ar}$ values were calculated from the measured O₂ concentrations, the O₂/Ar ratios, the observed SST and salinities (SSS) from the ship's thermosalinograph and the local barometric pressure. For O₂, the solubility parameterization of Garcia and Gordon [1992] based on data from Benson and Krause [1984] is used. For Ar, we use the parameterization of Hamme and Emerson [2004]. SST measured during the CTD cast was within $(0.00 \pm 0.04)^\circ\text{C}$ of the temperature at the seawater intake, but SSS had to be corrected upward by 0.20 psu, due to calibration errors of the thermosalinograph. The argon supersaturation (ΔAr) can be inferred from ΔO_2 and $\Delta O_2/\text{Ar}$ as $\Delta\text{Ar} = (\Delta O_2 - \Delta O_2/\text{Ar})/(1 + \Delta O_2/\text{Ar})$.

3. Results

[14] This cruise was the first deployment of our newly-developed instrument. Due to experimental problems with corrosion and clogged membranes, O₂/Ar ratio coverage is not continuous. We obtained reliable O₂/Ar data from the zonal transects and from two partial meridional transects between 8°S and the equator at 110°W and between 8°N and the equator at 95°W.

[15] Continuous ΔO_2 records based on optode data are generally similar between 2°N and 8°N for the 95°W and 110°W transects (Figure 1). Clear differences between 95°W and 110°W are seen in the upwelling region near the equator and smaller differences exist south of the equator, although the ΔO_2 variations at 95°W and 110°W are qualitatively similar between 2°S and 8°S. The surface water ΔO_2 and SST variations reflect the physical transport mechanisms that prevail in the equatorial Pacific. Foremost, the upwelling of cold, O₂-undersaturated waters is evident between the equator and 2°S at 95°W and between just south of 2°N and 1.5°S at 110°W. The northern transition corresponds to the boundary between the westward flowing South Equatorial Current (SEC) and the eastward flowing North Equatorial Countercurrent (NECC). It is particularly

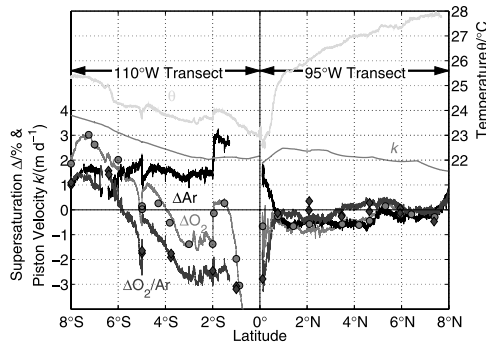


Figure 2. O₂, Ar and O₂/Ar supersaturation (ΔO_2 , ΔAr , $\Delta O_2/Ar$), temperature (θ) and piston velocity (k) during the southern part of the 110°W transect and the northern part of the second 95°W transect. Filled circles correspond to Winkler O₂ measurements, filled diamonds to discrete O₂/Ar ratio measurements. See color version of this figure in the HTML.

well-articulated at 2°N near 110°W and occurs within less than half a nautical mile. South of the equator, waters are re-saturated at about 2°S at 95°W. However, at 110°W O₂ is again undersaturated between 2°S and 4°S. This could be a remnant due to lateral advection of low O₂ waters from the coastal upwelling region west of Peru [Toggweiler *et al.*, 1991]. Another upwelling region is located between 9°N and 10°N near the boundary between NECC and the westward flowing North Equatorial Current.

[16] Outside of the regions influenced by upwelled waters, O₂ is at or above its saturation concentration, except for a region between 5°N and 1°N sampled during the second 95°W transect. A possible explanation for this undersaturation is to be found in the equally undersaturated Ar concentrations with ΔAr values down to -0.8% . There are two mechanisms that could explain its origin: a recent increase in barometric pressure (which contributed about -0.2%) or a decrease in temperature (which was not observed for the time between the first and second 8°N to 0°-transect at 95°W). Sampling and analysis errors may also contribute. This is suggested by a duplicate discrete $\Delta O_2/Ar$ measurement at 2°N, which is 0.5% higher than the $\Delta O_2/Ar$ values measured by MIMS. South of the equator, ΔAr values range from 1 to 3% (Figure 2), once again confirming the importance of correcting for the physical supersaturation of O₂.

[17] Figure 2 shows the $\Delta O_2/Ar$ data for the northern part of the first 95°W transect and the southern part of the 110°W transect, in addition to ΔO_2 optode measurements, sea surface temperatures and inferred ΔAr values. Both the MIMS and optode give excellent spatiotemporal resolution.

[18] As discussed above, $\Delta O_2/Ar$ can be used to estimate the contribution of biology to the observed O₂ supersaturation. Neglecting vertical mixing, a mixed-layer O₂ balance can be written as

$$z_{\text{mix}} \frac{\partial c(O_2)}{\partial t} = G - R - k(O_2)c_{\text{sat}}(O_2)\Delta O_2 + F_{\text{inj}}\chi(O_2) + F_{\text{exch}}\alpha(O_2)\chi(O_2)/\sqrt{Sc(O_2)}, \quad (1)$$

where z_{mix} is mixed layer depth, G is gross production, R is respiration, $k(O_2)$ is the O₂ gas-exchange coefficient (or

piston velocity), $c_{\text{sat}}(O_2)$ is the O₂ saturation concentration, and $F_{\text{inj}}/F_{\text{exch}}$ are air-sea fluxes due to bubble injection and bubble exchange, with the dimensionless parameters being $\chi(O_2)$ – O₂ mixing ratio in air, $Sc(O_2)$ – Schmidt number of O₂, $\alpha(O_2)$ – Ostwald solubility coefficient ([Hamme and Emerson, 2002] replace the diffusion coefficient of Hamme and Emerson's equation (4) by the Schmidt number for consistency). Taking the time-derivative of (1) and assuming $\partial^2 c/\partial t^2 = 0$ and constant z_{mix} , G , R , $k(O_2)$, and F_{bubble} , we obtain

$$k(O_2)c_{\text{sat}}(O_2)\Delta O_2 = G - R + F_{\text{inj}}\chi(O_2) + F_{\text{exch}}\alpha(O_2)\chi(O_2) / \sqrt{Sc(O_2)} - z_{\text{mix}} \sum_i \frac{\partial c_{\text{sat}}(O_2)}{\partial \psi_i} \frac{\partial \psi_i}{\partial t}. \quad (2)$$

ψ represents the influence of temperature, salinity and pressure changes on c_{sat} . A similar equation applies to Ar, with $G = R = 0$. Assuming the same relative response of Ar and O₂ to variations in ψ_i (true to within 1% for seawater) yields $F_{\text{bio}} = G - R =$

$$k(O_2)c_{\text{sat}}(O_2) \left[\Delta O_2 - \frac{k(Ar)}{k(O_2)} \Delta Ar \right] - F_{\text{inj}}\chi(O_2) \left[1 - \frac{c_{\text{sat}}(O_2)\chi(Ar)}{c_{\text{sat}}(Ar)\chi(O_2)} \right] - F_{\text{exch}}\alpha(O_2)\chi(O_2)/\sqrt{Sc(O_2)} \left[1 - \sqrt{\frac{Sc(O_2)}{Sc(Ar)}} \right] \quad (3)$$

[19] F_{bio} denotes biologically induced sea-air O₂ fluxes, which can be equated with mixed-layer NCP in areas where the above assumptions are valid and there is negligible vertical mixing across the base of the mixed layer. F_{bio} , G , R , k , $c_{\text{sat}}(O_2)$, and $F_{\text{bubble}}\chi(O_2)$ are O₂ fluxes in units of $\text{mmol m}^{-2} \text{d}^{-1}$. To calculate k , 24 h-average winds at the 10 m-height level (NCEP Reanalysis Project, <http://www.cdc.noaa.gov/>) are used in combination with the quadratic wind-speed gas-exchange relationship and Schmidt numbers of Wanninkhof [1992]. The first bracketed term in (3) is equal to $\Delta O_2/Ar + (\Delta O_2/Ar - 0.01)\Delta Ar$ and can be approximated by $\Delta O_2/Ar$. The second/third bracketed terms are about 0.1/ -0.01 , and will be neglected, because even if the entire observed Ar supersaturation of mostly $<2\%$ were due to bubbles, the O₂/Ar supersaturation would be affected by $<0.2\%$. F_{bio} is therefore approximated by $k(O_2)c_{\text{sat}}(O_2)\Delta O_2/Ar$. Values for k are calculated iteratively from the fraction of the mixed layer ventilated on a particular day before sampling, for 120 days backwards in time (M. K. Reuer *et al.*, New estimates of Southern Ocean biological production rates from O₂/Ar ratios and the triple oxygen isotope composition of O₂, submitted to *Deep-Sea Research, Part I*, 2005). Mixed layer depths, taken here as the depth where θ is 0.5°C less than at the surface, vary from 10 m in upwelling regions near the equator at 10°N to 93 m at 2.5°S on the 110°W line. Mixed layer depths only enter implicitly into the calculation.

[20] The derived F_{bio} values are shown in Figures 3 and 4 for the meridional and zonal transects, respectively. Clearly, F_{bio} is higher at 8°S ($5\text{--}18 \text{ mmol m}^{-2} \text{d}^{-1}$) than at 8°N ($0\text{--}4 \text{ mmol m}^{-2} \text{d}^{-1}$). To the extent that F_{bio} can be related to NCP, this meridional gradient is in semi-quantitative agree-

ment with nitrate-based estimates of new production in the eastern equatorial Pacific [Fiedler *et al.*, 1991], which showed about 10 times higher new production at 8°S than at 8°N. This may be due to higher nutrient concentrations in the south, advected into the region by the SEC from the Peru upwelling area, which is supported by modeling results [Jiang and Chai, 2005]. Nitrate concentrations were indeed between 6 and 9 mmol m⁻³ south of the equator, and below 0.4 mmol m⁻³ north of 0.5°N at 95°W and north of 2°N at 110°W, suggesting nitrate limitation in the north. Silicate was shown to regulate new production further west in the equatorial Pacific (140°W) [Dugdale and Wilkerson, 1998], but was not measured during the present cruise. Low iron concentrations, together with data showing that iron amendments enhance growth [Gordon *et al.*, 1998], indicate that the waters are iron-limited [Coale *et al.*, 1996]. However, iron concentrations seem to be lower south of the equator than north of it, which would rather support a productivity trend opposite to the one we observed [Jiang and Chai, 2005].

4. Discussion and Conclusions

[21] Our simple O₂ mass balance approach does not constrain net community production (NCP) where vertical mixing affects the O₂ budget, especially in the equatorial upwelling region. Similar to Hendricks *et al.* [2005] and based on the observed SST and ΔO₂ values, we therefore restrict our analysis to north of 2.75°N and south of 6.75°S. North of the equator, mixed-layer NCP is low throughout a reach of 600 km between 8°N to 2.75°N. South of the equator, we cannot say how far the higher productivities extend north from 6.75°S because of the competing influences of productivity, upwelling/advection of low O₂ waters and gas exchange. Our NCP estimates are similar to previous values derived from discrete O₂/Ar measurements, nitrate budgets and ¹⁵N incubations [Hendricks *et al.*, 2005, and references therein].

[22] O₂/Ar ratios, together with wind-speed gas-exchange parameterizations, yield the biologically induced sea-air O₂ flux, F_{bio} . This term integrates NCP over larger reaches of the ocean when vertical mixing and lateral mixing are neglected, but an exhaustive interpretation likely requires an ocean transport model. The potential of MIMS to measure other gases and gas ratios [Tortell, 2005] and to provide absolute gas concentrations in connection with

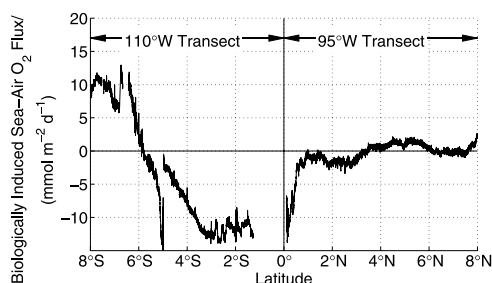


Figure 3. Biologically induced sea-air O₂ fluxes, F_{bio} , inferred from ΔO₂/Ar observations and piston velocities calculated from NCEP 10 m-winds and the wind-speed gas-exchange relationship of Wanninkhof [1992].

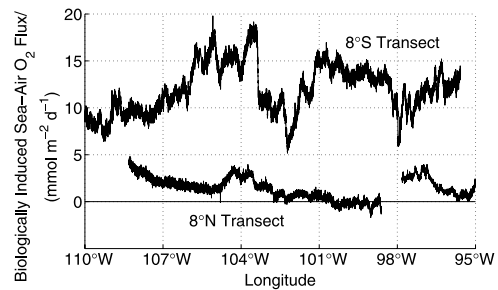


Figure 4. Zonal variations in F_{bio} along the 8°S and 8°N transects.

precise O₂ concentrations measurements make it a very promising technique, e.g., in order to use of ΔAr values to infer heat fluxes or ΔN₂ values to unravel the influence of bubble injection.

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